

(η^5 -4,7-Dimethylindenyl)(η^5 -pentamethylcyclopentadienyl)iron(II)**Peter G. Jones,^{a*} Henning Hopf^b and Thorsten Hartig^b**^aInstitut für Anorganische und Analytische Chemie, Technische Universität Braunschweig, Postfach 3329, 38023 Braunschweig, Germany, and ^bInstitut für Organische Chemie, Technische Universität Braunschweig, Postfach 3329, 38023 Braunschweig, Germany

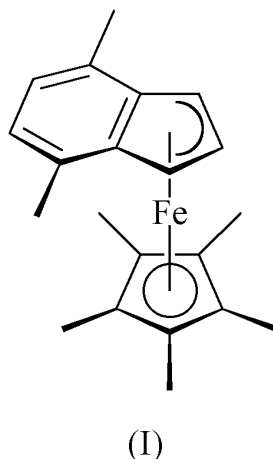
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Key indicatorsSingle-crystal X-ray study
 $T = 178\text{ K}$
Mean $\sigma(\text{C}-\text{C}) = 0.004\text{ \AA}$
 R factor = 0.036
 wR factor = 0.089
Data-to-parameter ratio = 13.4For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The molecule of the title compound, $[\text{Fe}(\text{C}_{10}\text{H}_{15})(\text{C}_{11}\text{H}_{11})]$, contains parallel and eclipsed five-membered rings coordinated to the central Fe atom at perpendicular distances of 1.642 (1) (cyclopentadienyl ring) and 1.667 (1) Å (indenyl ring).

Comment

In our studies of the preparation and uses of novel metallocenes containing the [2.2]paracyclophane unit (Hopf *et al.*, 2002), we needed the title ferrocene, (I), which incorporates a 'half-cyclophane' as a subunit, as a reference compound. Here, we present its structure.



The molecule is shown in Fig. 1. The five-membered rings are planar (mean deviation 0.008 Å for C1 to C7A and 0.002 Å for C15 to C19), essentially parallel [interplanar angle 2.8 (2)°] and almost ideally eclipsed [torsion angle *e.g.* C2—Cent1—Cent2—C16 −0.3 (2)°, where Cent represents the centroid of the ring C atoms]. The perpendicular distance of the Fe atom from the rings is 1.667 (1) and 1.642 (1) Å, respectively. The coordination to the indenyl ring is very slightly distorted towards η^3 , the distances Fe—C3A and Fe—C7A being significantly longer than the others (Table 1).

The molecules pack (Fig. 2) in layers parallel to the *ab* plane, with a herring-bone pattern.

Experimental

The title compound was prepared in 85% yield (Hartig, 1991) by first metallating 4,7-dimethyl-1*H*-indene with methyl lithium in anhydrous tetrahydrofuran and then quenching the anion thus formed with an equivalent amount of the complex $[(\text{C}_5\text{Me}_5)\text{Fe}(\text{CH}_3\text{CN})_3]\text{PF}_6$. Crystals were grown by cooling of a hot saturated solution in hexane.

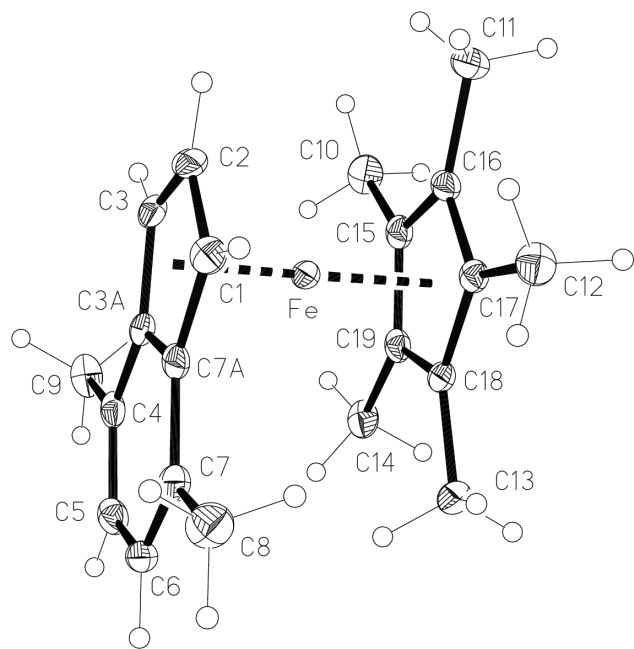


Figure 1
The molecule of the title compound in the crystal. Ellipsoids are drawn at the 30% probability level.

Crystal data

[Fe(C₁₀H₁₅)(C₁₁H₁₁)]

$M_r = 334.27$

Monoclinic, $P2_1/n$

$a = 8.350(2) \text{ \AA}$

$b = 13.925(3) \text{ \AA}$

$c = 14.645(4) \text{ \AA}$

$\beta = 95.88(2)^\circ$

$V = 1693.9(7) \text{ \AA}^3$

$Z = 4$

$D_x = 1.311 \text{ Mg m}^{-3}$

Mo $K\alpha$ radiation

Cell parameters from 48 reflections

$\theta = 10\text{--}12^\circ$

$\mu = 0.89 \text{ mm}^{-1}$

$T = 178(2) \text{ K}$

Tablet, red

$0.60 \times 0.40 \times 0.15 \text{ mm}$

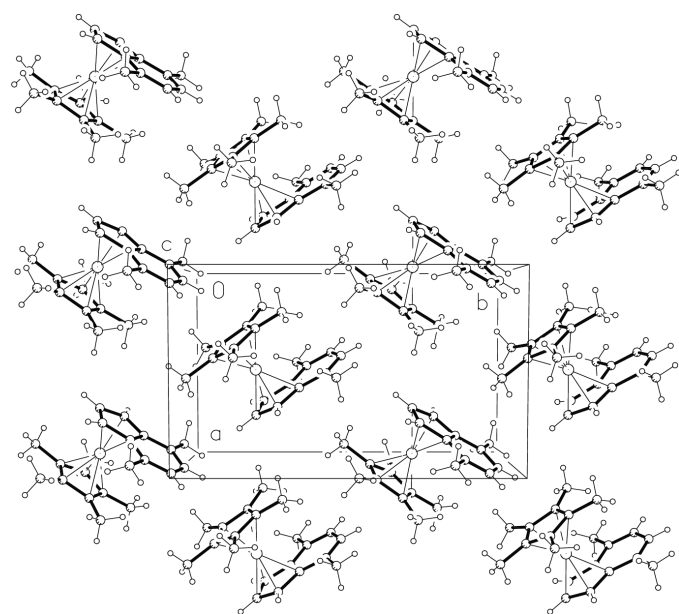


Figure 2
Packing diagram of the title compound with view direction parallel to the c axis. There are two such layers, related by inversion symmetry, per c axis repeat. Radii are arbitrary.

Data collection

Nicolet R3 diffractometer
 ω scans
Absorption correction: none
2821 measured reflections
2768 independent reflections
2191 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.020$

$\theta_{\text{max}} = 25.0^\circ$
 $h = -9 \rightarrow 9$
 $k = -16 \rightarrow 0$
 $l = -17 \rightarrow 17$
3 standard reflections
every 247 reflections
intensity decay: none

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.036$
 $wR(F^2) = 0.089$
 $S = 1.04$
2768 reflections
206 parameters
H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.0459P)^2 + 0.6303P]$
where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.37 \text{ e \AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.46 \text{ e \AA}^{-3}$

Table 1

Selected interatomic distances (\AA).

Fe—C16	2.037 (3)	C3—C3A	1.434 (4)
Fe—C2	2.039 (2)	C3A—C4	1.429 (4)
Fe—C15	2.040 (3)	C3A—C7A	1.440 (4)
Fe—C17	2.044 (3)	C4—C5	1.356 (4)
Fe—C3	2.045 (3)	C5—C6	1.427 (4)
Fe—C18	2.046 (2)	C6—C7	1.360 (4)
Fe—C19	2.049 (2)	C7—C7A	1.431 (4)
Fe—C1	2.049 (3)	C15—C19	1.432 (4)
Fe—C3A	2.089 (3)	C15—C16	1.433 (4)
Fe—C7A	2.093 (3)	C16—C17	1.432 (4)
C1—C2	1.414 (4)	C17—C18	1.425 (3)
C1—C7A	1.427 (4)	C18—C19	1.425 (4)
C2—C3	1.423 (4)		

H atoms at sp^2 C atoms were included using a riding model, starting from idealized positions. Methyl H atoms were located in difference syntheses, idealized and refined as rigid groups allowed to rotate but not tip.

Data collection: *P3* (Nicolet, 1987); cell refinement: *P3*; data reduction: *XDISK* (Nicolet, 1987); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1990); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *XP* (Siemens, 1994); software used to prepare material for publication: *SHELXL97*.

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References

- Hartig, T. (1991). PhD thesis, Technical University of Braunschweig, Germany.
Hopf, H., Sankararaman, S., Dix, I., Jones, P. G., Alt, H. G. & Licht, A. (2002). *Eur. J. Inorg. Chem.* pp. 123–131, and references therein.
Nicolet (1987). *P3* and *XDISK*. Nicolet Instrument Corporation, Madison, Wisconsin, USA.
Sheldrick, G. M. (1990). *Acta Cryst.* **A46**, 467–473.
Sheldrick, G. M. (1997). *SHELXL97*. University of Göttingen, Germany.
Siemens (1994). *XP*. Version 5.03. Siemens Analytical X-ray Instruments, Madison, USA.